

Deformylative Intramolecular Hydroarylation: Synthesis of Benzo[e]pyrido[1,2-a]indoles

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S Supporting Information

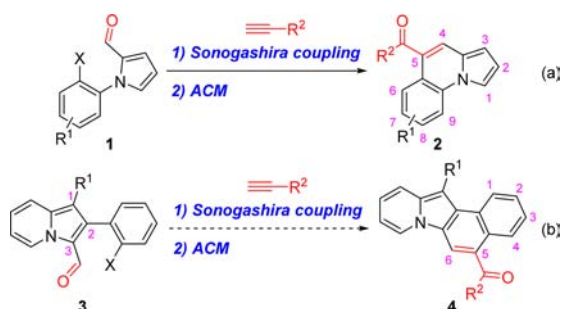
ABSTRACT: Attempted cyclization of indolizines bearing both formyl and alkyne groups under acid catalysis provided benzo[e]pyrido[1,2-a]indoles with an aryl substituent at the C6 position as major products, along with the expected C5-acylated benzo[e]pyrido[1,2-a]indoles as minor ones, which resulted from preferential deformylative intramolecular hydroarylation instead of intended alkyne-carbonyl metathesis.



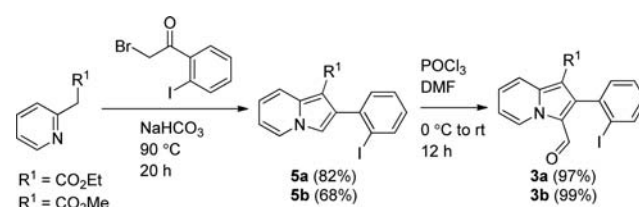
As part of our interest in the design and synthesis of indolizine-embedded skeletons,¹ we recently communicated the construction of pyrrolo[1,2-a]quinolines with an acyl group at the C5 position by way of successive Sonogashira cross-coupling/intramolecular alkyne-carbonyl metathesis (ACM) (Scheme 1a).² Along these lines, we surmised that 2-(2-haloaryl)indolizine-3-carbaldehydes such as **3** would be converted to the corresponding C5-acyl benzo[e]pyrido[1,2-a]indoles by similar procedures (Scheme 1b).

Despite its presence in many biologically active natural products, pharmaceuticals, and imaging agents,³ not many synthetic approaches to this chemical core structure have appeared in the literature.⁴ Although this tetracyclic compound was accessed by employing Larock's cascade carbopalladation–annulation protocol,^{4a} regioselective incorporation of unsymmetrical internal alkynes was not realized.⁵ As our strategy allows for not only the regioselective introduction of a substituent but also a unique substitution pattern around this polycyclic structure, we decided to investigate an intramolecular ACM route to benzo[e]pyrido[1,2-a]indoles decorated with an acyl moiety at the C5 site. During our study, we discovered that an unprecedented deformylative intramolecular hydroarylation⁶ was a major pathway leading to formation of benzo[e]pyrido-

Scheme 1. Synthetic Strategies to Indolizine-Containing Structures via Sonogashira Coupling/Intramolecular ACM

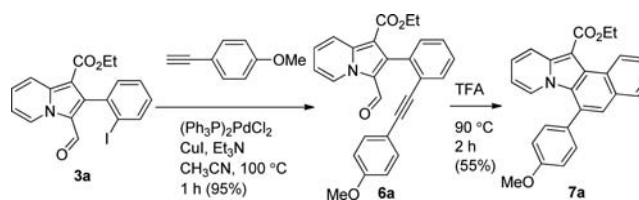


Scheme 2. Synthesis of 3^a



^aA mixture of 2-bromo-2'-iodoacetophenone (3.45 mmol), ethyl or methyl 2-pyridylacetates (5.18 mmol, 1.5 equiv) and NaHCO₃ (2.0 equiv) in acetone (10 mL) was heated at 90 °C. A mixture of **5** (2.04 mmol) and POCl₃ (3 equiv) in DMF (1 mL) was stirred at 0 °C to rt.

Scheme 3. Unexpected Formation of 7a^a



^aA mixture of **3a** (1.19 mmol), 4-ethynylanisole (2 equiv), (Ph₃P)₂PdCl₂ (0.1 equiv), CuI (0.1 equiv), and Et₃N (4 equiv) in CH₃CN (25 mL) was heated at 100 °C. A mixture of **6a** (0.07 mmol) in TFA (0.6 mL) was stirred at 90 °C.

[1,2-a]indoles with an aryl or alkyl group at the C6 position, which we wish to report here.

This study began with preparation of the requisite 2-(2-iodophenyl)indolizine-3-carbaldehydes **3** (Scheme 2). Base-mediated condensation of 2-pyridylacetates with 2-bromo-2'-iodoacetophenone provided indolizines **5** in good yields.

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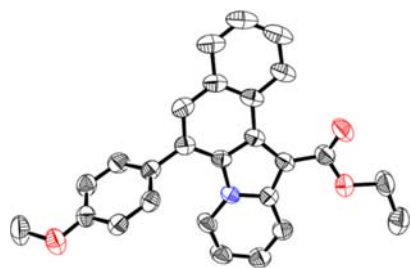
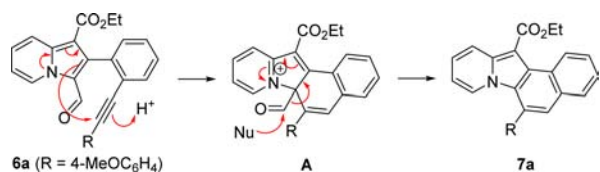
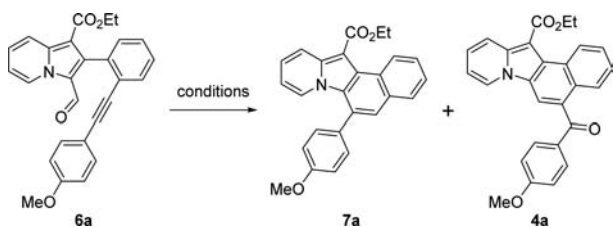


Figure 1. Crystal structure of 7a.

Scheme 4. Proposed Mechanism for Formation of 7a

Table 1. Reaction of 6a under Various Conditions^a

entry	catalyst (equiv)	temp (°C)	time (h)	solvent	yield (%) ^b	
					7a	4a
1	TFA	rt	1	–	80	13
2	TFA	rt	1	DCE	72	18
3	InCl ₃ (0.3)	rt to 60	72	DCE	72	6
4	In(OTf) ₃ (0.3)	rt	48	DCE	56	33
5	In(OTf) ₃ (0.05)	60	22	DCE	60	21
6	In(OTf) ₃ (0.05)	60	28	DCM	60	33

^aA mixture of 6a (0.07 mmol) and catalyst in solvent (1 mL) was stirred at the indicated temperature and time unless otherwise noted. ^bIsolated yield (%).

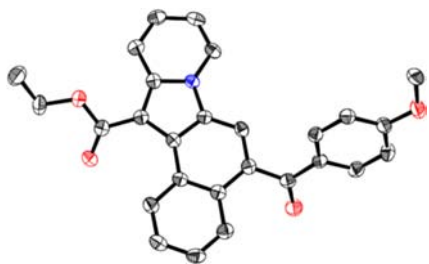
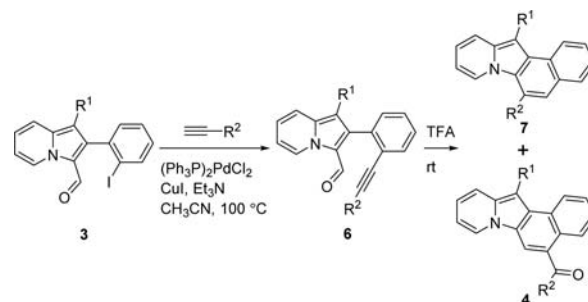


Figure 2. Crystal structure of 4a.

Subsequent Vilsmeier–Haack formylation took place uneventfully to afford the corresponding aldehydes 3.

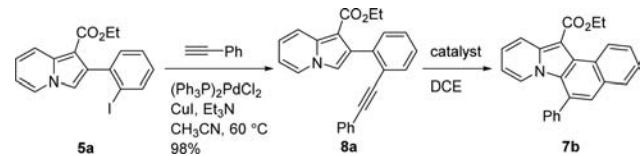
At this point, we used 3a to optimize the following reaction conditions. Sonogashira cross-coupling of 3a with 4-ethynylanisole furnished 6a in excellent yield (Scheme 3). Based on our previous results, 6a was exposed to TFA at 90 °C to induce intramolecular ACM. Surprisingly, however, the major product was not the expected 4a; instead, 7a was isolated in 55% yield. The structure of 7a was unambiguously established by X-ray crystallographic analysis (Figure 1).⁷ Although deformylation of

Table 2. Syntheses of 7 and 4 from 3^a

entry	3	6 (%) ^b	7 (%) ^b	4 (%) ^b
1		6b (R ² = Ph) (92)	7b (47)	trace
2	3a	6c (R ² = 4-MeC ₆ H ₄) (90)	7c (76)	4c (13)
3	3a	6d (R ² = 4-MeO-2-MeC ₆ H ₃) (89)	7d (86)	trace
4	3a	6e (R ² = 4-NCCH ₂ C ₆ H ₄) (76)	7e (48)	trace
5	3a	6f (R ² = 6-MeO-2-naphthalene) (78)	7f (79)	4f (15)
6	3a	6g (R ² = 1-naphthalene) (92)	7g (63)	4g (7)
7	3a	6h (R ² = cyclopentyl) (90)	c	
8	3a	6i (R ² = n-Bu) (36)	c	
9		6j (R ² = Ph) (94)	7j (56)	trace
10	3b	6k (R ² = 3-MeOC ₆ H ₄) (95)	7k (72)	trace
11	3b	6l (R ² = 4-PhC ₆ H ₄) (88)	7l (60)	4l (14)
12	3b	6m (R ² = 3,5-(MeO) ₂ C ₆ H ₃) (98)	7m (47)	trace
13	3b	6n (R ² = 3-thiophene) (99)	7n (50)	4n (22)

^aA mixture of 3 (0.24 mmol), alkyne (2 equiv), (Ph₃P)₂PdCl₂ (0.1 equiv), CuI (0.1 equiv), and Et₃N (4 equiv) in CH₃CN (2 mL) was heated at 100 °C. A mixture of 6 (0.07 mmol) in TFA (0.6 mL) was stirred at rt. ^bIsolated yield (%) ^cComplex mixture.

compounds possessing pyrrole-2-carboxaldehyde under acidic conditions was known in the literature,⁸ to the best of our knowledge, no examples where intramolecular cyclization occurs at the site bearing a formyl group with concomitant deformylation have been disclosed. This is surprising because the compound derived from 1 was transformed into 2 in good yield without loss of a formyl group under similar conditions. As shown in Scheme 4, formation of 7a can be understood by

Table 3. Synthesis of 7 from 8a^a

entry	catalyst (equiv)	temp (°C)	time (h)	7b (%) ^b
1	TFA ^c	rt to 60	24	50
2	In(OTf) ₃ (0.1 equiv)	rt to 40	24	42
3	Bi(OTf) ₃ (0.1 equiv)	rt to 40	24	36
4	AgOTf (0.2 equiv)	rt	5	64
5	TfOH (0.2 equiv)	0 to rt	24	40 (53) ^d

^aA mixture of 5a (0.767 mmol), alkyne (2 equiv), (Ph₃P)₂PdCl₂ (0.1 equiv), CuI (0.1 equiv), and Et₃N (4 equiv) in CH₃CN (6 mL) was heated at 60 °C. A mixture of 8a (0.03 mmol) and catalyst (0.2 equiv) in DCE (0.5 mL) was stirred at the indicated temperature unless otherwise stated. ^bIsolated yield (%) ^cTFA was used as solvent. ^dYield in parentheses is based on recovered starting material.

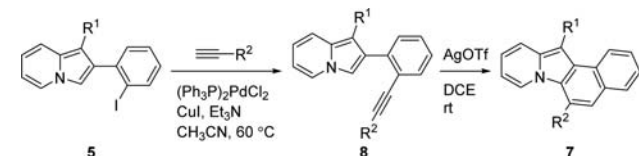
initial attack of the C3 site of indolizine 6a to the neighboring alkyne unit to form the indolizinium salt A, which is followed by deformylation to restore aromaticity.⁹

Lowering the reaction temperature to rt produced 7a as well as 4a (Table 1, entry 1). The structure of 4a was again confirmed by X-ray crystal analysis as shown in Figure 2.¹⁰ Reaction in TFA/DCE mixture gave a similar result (entry 2). In(OTf)₃ exhibited superior reactivity to InCl₃ (entries 3 and 4). Exposure of 6a to In(OTf)₃ (0.05 equiv) at 60 °C led to a mixture of products in a ratio of 3:1 or 2:1 (entries 5 and 6). It seemed that deformylative intramolecular hydroarylation was a preferred pathway rather than the ACM we initially anticipated.

When several indolizines 6 having other alkynes were prepared and treated with TFA at rt, similar results were observed (Table 2). However, when R² is cyclopentyl or *n*-butyl, a complex mixture of products was observed (entries 7 and 8).

As compound 7 could be directly obtained via the intramolecular hydroarylation of 8, we synthesized 8 to investigate the reactivity of these compounds toward 6-endo-dig cyclization (Table 3). To this end, Sonogashira coupling of 5a was first conducted at 60 °C to give the corresponding adduct 8a in good yields. When 8a was treated under the same conditions applied for 6, surprisingly, no reaction took place. The cyclized product was obtained in 50% yield by increasing the temperature to 60 °C (entry 1). Although the isolated product yields are about the same, the cyclization reactivity of 6a seemed higher than that of 8a, supporting the idea that transformation of 6 to 7 does not occur via 8. Use of In(OTf)₃ or Bi(OTf)₃ as catalysts at 40 °C for 24 h produced the desired 7b in 42% and 36% yield, respectively (entries 2 and 3). Reaction of 8a under the influence of AgOTf provided 7b in 64% yield (entry 4). Incomplete conversion was observed with triflic acid even after 24 h (entry 5).

To examine the generality of this intramolecular hydroarylation, 5 was reacted with more alkynes to afford the various adducts in good to excellent yields (Table 4). Overall, the subsequent Ag(I)-catalyzed hydroarylation¹¹ went smoothly with arylalkyne-substituted indolizines. Modest yields of the products were obtained with substrates having an alkylalkyne (entries 7 and 8). Thiophene-containing benzo[e]pyrido[1,2-*a*]indole 7n was isolated in 74% yield (entry 14).

Table 4. Syntheses of 7 via Ag(I)-Catalyzed Hydroarylation^a

entry	5	8 (%) ^b	7 (%) ^b
1		8a (R ² = Ph) (98)	7b (64)
2	5a	8b (R ² = 4-MeOC ₆ H ₄) (91)	7a (87)
3	5a	8c (R ² = 4-MeC ₆ H ₄) (96)	7c (70)
4	5a	8d (R ² = 4-MeO-2-MeC ₆ H ₃) (88)	7d (94)
5	5a	8e (R ² = 4-NCCH ₂ C ₆ H ₄) (98)	7e (68)
6	5a	8f (R ² = 6-MeO-2-naphthalene) (80)	7f (79)
7	5a	8g (R ² = 1-naphthalene) (92)	7g (74)
8	5a	8h (R ² = cyclopentyl) (92)	7h (29) ^c
9	5a	8i (R ² = <i>n</i> -Bu) (88)	7i (40) ^c
10		8j (R ² = Ph) (96)	7j (67)
11	5b	8k (R ² = 3-MeOC ₆ H ₄) (74)	7k (87)
12	5b	8l (R ² = 4-PhC ₆ H ₄) (96)	7l (79)
13	5b	8m (R ² = 3,5-(MeO) ₂ C ₆ H ₃) (85)	7m (78)
14	5b	8n (R ² = 3-thiophene) (97)	7n (74)
15	5b	8o (R ² = 6-MeO-2-naphthalene) (79)	7o (86)

^aA mixture of 5 (0.767 mmol), alkyne (2.0 equiv), (Ph₃P)₂PdCl₂ (0.1 equiv), CuI (0.1 equiv), and Et₃N (4.0 equiv) in CH₃CN (6 mL) was heated at 60 °C. A mixture of 8 (0.08 mmol) and AgOTf (0.2 equiv) in DCE (1 mL) was stirred at rt. ^bIsolated yield (%) ^cReaction temperature: rt to 60 °C

In summary, during our endeavor to construct benzo[*e*]pyrido[1,2-*a*]indoles with an acyl moiety at the C5 site via intramolecular alkyne-carbonyl metathesis, an unprecedented intramolecular hydroarylation with concomitant extrusion of a formyl group was discovered to give rise to the same skeleton with an aryl substituent at the C6 position as the major product under acidic media. Alternatively, facile formation of the latter from the indolizines without a formyl group at the C3 position was demonstrated in the presence of a Ag(I) catalyst. Comparison of the reactivity of both substrates revealed that the presence of a formyl group at the C3 position of indolizines

seems to facilitate intramolecular hydroarylation. A plausible mechanism is proposed. These findings will enable benzo[e]-pyrido[1,2-*a*]indoles to be synthesized with suitable substituents. More efforts to decorate this skeleton in a regioselective manner as well as evaluation of these new compounds in various settings are underway.

■ ASSOCIATED CONTENT

■ Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: [10.1021/acs.orglett.5b02331](https://doi.org/10.1021/acs.orglett.5b02331).

File for 7a (CIF)

File for 4a (CIF)

Full experimental details, ¹H and ¹³C NMR spectra of 3–8 (PDF)

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Notes

The authors declare no competing financial interest.

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(7) CCDC 1410649 contains the supplementary crystallographic data for compound 7a. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

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(9) Deformylated compounds such as **8** were not detected during TLC monitoring of the reaction. One reviewer commented that the aldehyde group could be protonated so that it may stabilize the resonance form pertaining negative charge on C3 carbon and positive charge on the indolizine nitrogen.

(10) CCDC 1410653 contains the supplementary crystallographic data for compound 4a. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

(11) For a recent review on Ag-catalyzed reactions of alkynes, see: Fang, G.; Bi, X. *Chem. Soc. Rev.* **2015**, DOI: [10.1039/C5CS00027K](https://doi.org/10.1039/C5CS00027K).